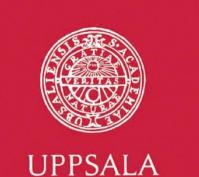


# **Assessment of Fuel Reabsorption by Thermally Pre-heated Co-deposited Layers**





**UNIVERSITET** 

D. Ivanova<sup>a</sup>, M. Rubel<sup>a</sup>, V. Philipps<sup>b</sup> B. Schweer<sup>b</sup>, M. Freisinger<sup>b</sup>, A. Schmidt<sup>b</sup> **P. Petersson**<sup>a,c</sup>

<sup>a</sup> Alfvén Laboratory, Royal Institute of Technology, Association EURATOM – VR, Stockholm, Sweden <sup>b</sup> Institute for Energy Research, Forschungszentrum Jülich, Association EURATOM, Jülich, Germany <sup>c</sup> Ångström Laboratory, Uppsala University, Association EURATOM – VR, Uppsala, Sweden

E-mail contact: darya.ivanova@ee.kth.se

### **Background and Motivation**

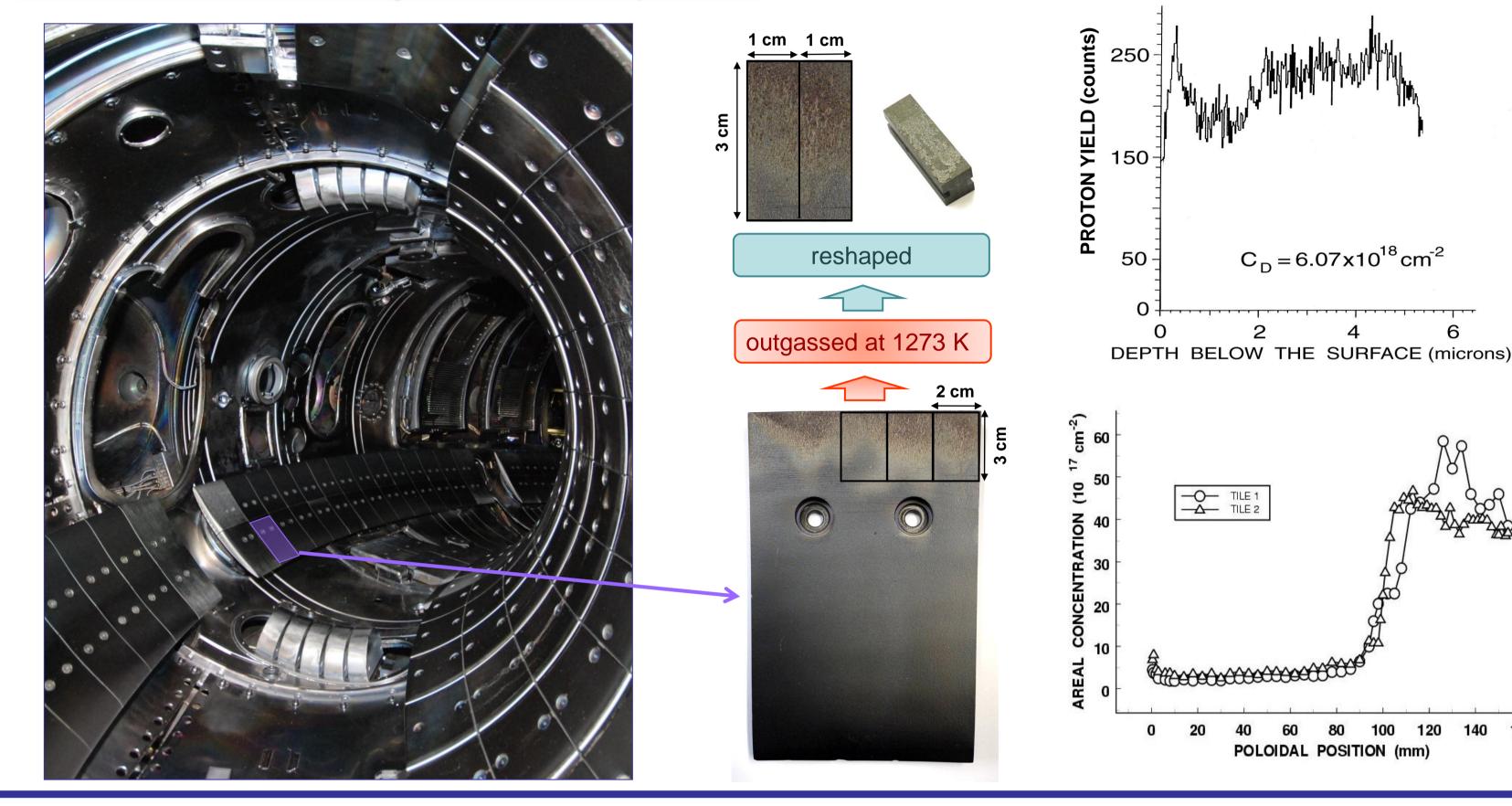
- Reduction of in-vessel fuel inventory in fusion devices is essential when operation with tritium is considered. Especially in presence of a carbon wall.
- Two basic schemes for fuel removal are currently considered: (i) desorption of hydrogen-containing species, (ii) removal of the entire fuel-rich co-deposit. In all cases, the outgassed deposited layers remain in the vessel and they would be repeatedly exposed to plasma.

□ How do the outgased layers respond to plasma during the repeated exposure?

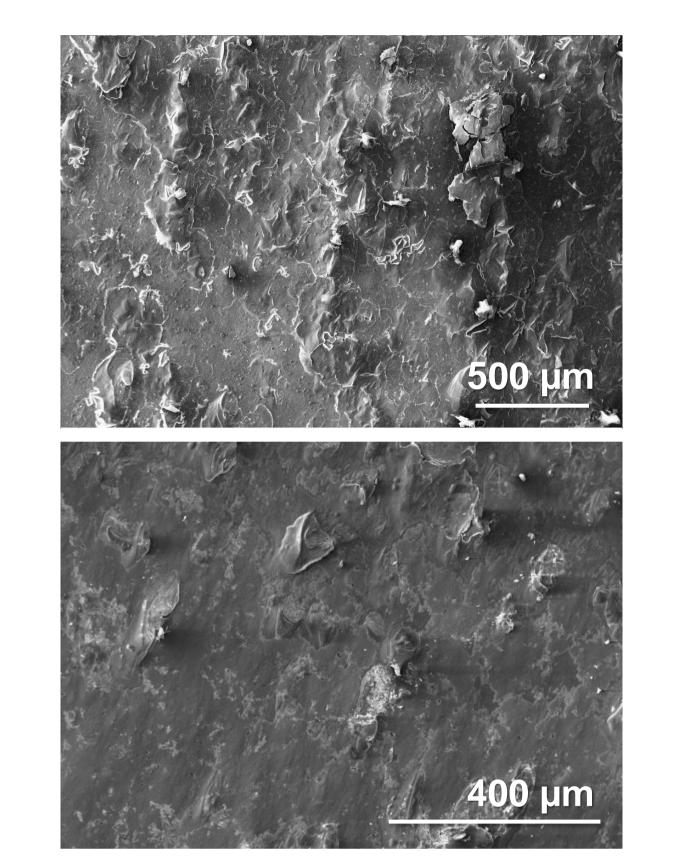
### **Analysis Methods**

- Scanning Electron Microscopy (SEM) with Energy Dispersive X-ray Spectroscopy (EDX)
- Thermal Desorption Spectrometry (TDS)
- Nuclear Reaction Analysis (NRA)
- Rutherford Backscattering Spectrometry (RBS)

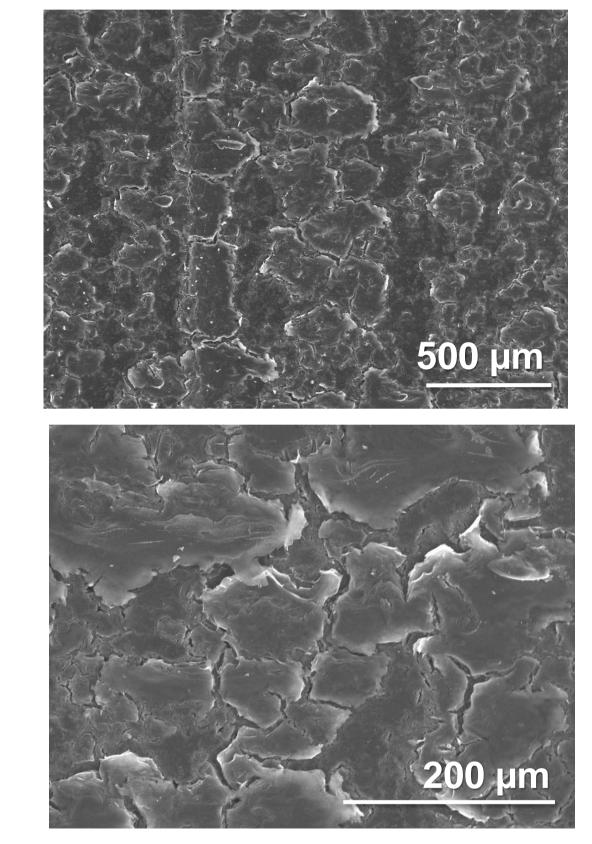
### Materials for study: ALT-II deposits



#### **Original deposits:**



**Deposits after outgassing:** 



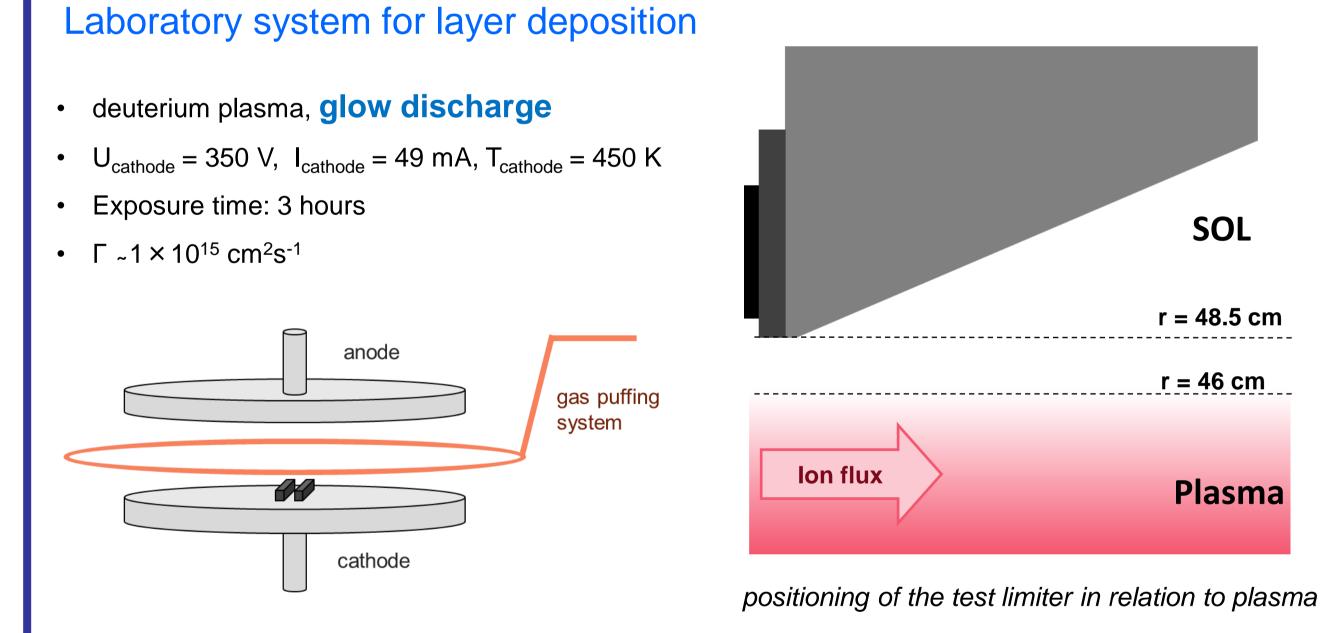
### **Experimental set-up**

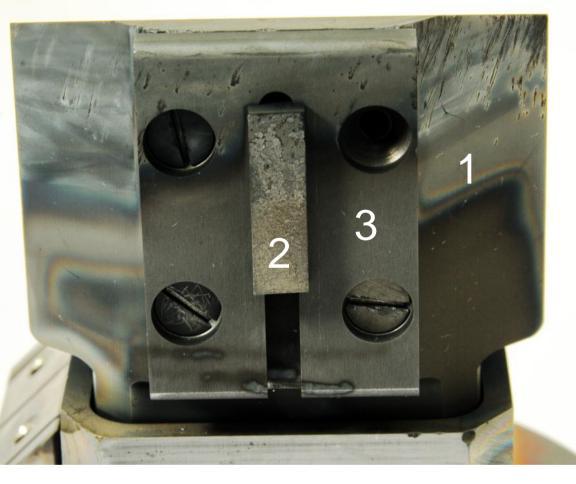
PADOS:

- TEXTOR **SOL plasma**, > 3 cm behind the main limiter
  - Plasma parameters:  $B_t = 2.2 2.6 \text{ T}$ ,  $n_e = 2.5 3 \times 10^{19} \text{ m}^{-3}$ ,  $I_p = 350 400 \text{ kA}$ 
    - 2 exposures: 40 and 25 plasma seconds
  - Γ~5×10<sup>19</sup> cm<sup>2</sup>s<sup>-1</sup>

### **Results: Nuclear Reaction Analysis**

2 MeV <sup>3</sup>He beam; D(<sup>3</sup>He,p)<sup>4</sup>He

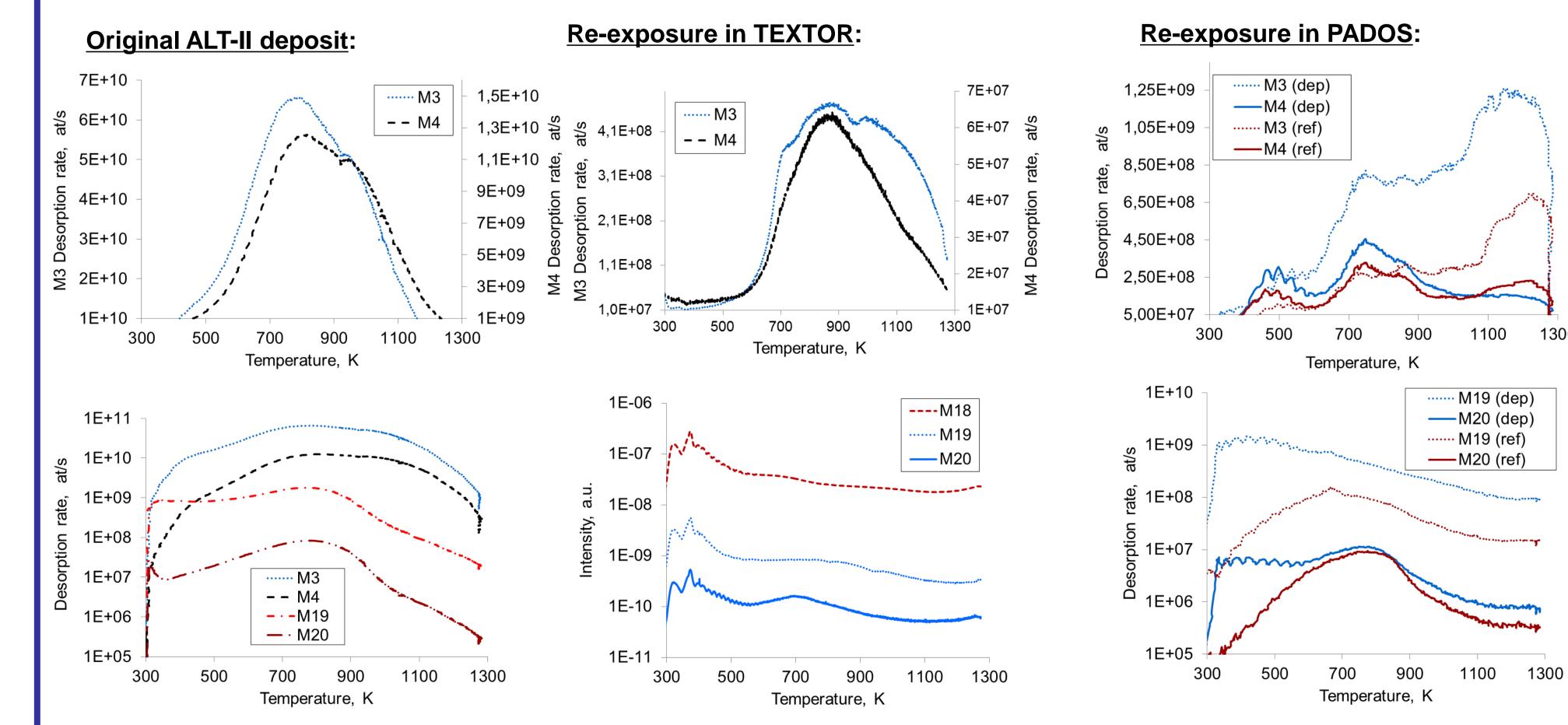




Test limiter after exposure in TEXTOR (1) with a specimen of ALT-II tile (2) and a pure graphite holder serving also as reference material (3).

## **Results: Thermal Desorption Spectrometry**

- Monitoring masses: M2 (H<sub>2</sub>), M3 (HD), M4 (D<sub>2</sub>), M19 (HDO+CHD<sub>3</sub>), M20 (D<sub>2</sub>O+CD<sub>4</sub>)
- Heating rate 0.055 K·s<sup>-1</sup>



			D retention (at∙cm⁻²)		remarks	
	Re-exposed deposit		4.7×10 <sup>18</sup>		broad depth distribution (> 7 μm)	
Pure graphite		phite	1.33×10 <sup>17</sup>		70% of fuel in the surface layer (1.5 $\mu$ m)	
	4,00E+17	• •	•	and re	deposit after outgassing -exposure in TEXTOR graphite plate (reference)	
n2]	3.00E+17	× ×				mm
ntent [at/cm2]	3,00E+17 2,00E+17		•			30 mm
D content [at/cm2]			◆ ★ ★	 		30 mm

Deuterium retention is low due to poor contact between

Position [mm]

a deposited layer and the substrate

10

# Summary

1300

Annealing at high temperatures (up to 1273 K) enhances layer

25

30

38 mm

brittleness leading eventually to detachment of co-deposits.

Deuterium depth distribution measured with NRA in re-exposed deposits is broad (6 µm) in comparison to the distribution in fresh graphite (< 1.5  $\mu$ m) used as reference material.

□ The measured values of the fuel retention in the re-exposed deposits are 30 to 40 times lower than in the pure graphite, showing that the fuel re-absorption does not lead to an immediate resaturation of deposits. This may be partly explained by the detachment of the original deposits during the outgassing procedure.

Desorption characteristics (M3, M4) for the original co-deposit and after repeated exposure in TEXTOR are the same, whereas exposure in PADOS results in several binding states..

□ All re-exposed deposits adsorb large amount of water vapor. One observes the presence of HDO and D<sub>2</sub>O probably attributed to isotope exchange.

- Water vapour is easily adsorbed by deposits and isotope exchange leads to HDO and  $D_2O$
- Temperature above 600 K are needed in order to remove hydrocarbons (CHD<sub>3</sub> and CD<sub>4</sub>) efficiently